Cyclocondensation of 6-Acetyl-4,7-dihydro-5-methyl-7-phenyl[1,2,4]triazolo[1,5-*a*]pyrimidine with Hydroxylamine and Hydrazine

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The cyclocondensation of 6-acetyl-4,7-dihydro-5-methyl-7-phenyl[1,2,4]triazolo[1,5-a]pyrimidine (3) with hydroxylamine or hydrazine leads to 3a,4,9,9a-tetrahydro-3,9a-dimethyl-4-phenylisoxazolo-[5,4-d][1,2,4]triazolo[1,5-a]pyrimidine (4a) and 3a,4,9,9a-tetrahydro-3,9a-dimethyl-4-phenyl-1H-pyrazolo[3,4-d][1,2,4]triazolo[1,5-a]pyrimidine (4b), respectively. In the presence of methanolic hydrogen chloride, 4b undergoes a cleavage of the pyrimidine ring to yield (5-amino-1,2,4-triazol-1-yl)(3,5-dimethylpyrazol-4-yl)phenylmethane (5). The structure determination of the compounds obtained is based on ¹H and ¹³C nmr spectra including NOE measurements.

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The reactions of acetyl-substituted dihydroazines with binucleophiles represent a convenient synthesis of partially hydrogenated azoloazines [1], compounds with interesting biological and pharmacological properties [2,3].

Now we report on the synthesis of new heterocyclic systems, namely substituted isoxazolo[5,4-d][1,2,4]triazolo[1,5-a]pyrimidine (4a) and 1H-pyrazolo[3,4-d]-[1,2,4]triazolo[1,5-a]pyrimidine (4b) by the cyclocondensation of 6-acetyl-4,7-dihydro-5-methyl-7-phenyl[1,2,4]triazolo[1,5-a]pyrimidine (3) with hydroxylamine or hydrazine.

Compound 3 was prepared by the reaction of 3-amino-1,2,4-triazole (1) with 2-benzylidenepentane-2,4-dione (2). Subsequent refluxing of 3 and hydroxylamine or hydrazine in methanol gave 4a and 4b, respectively. Compound 4a proved to be stable in deuteriochloroform which contained traces of deuterium chloride. In 4b the central ring was cleaved under these conditions. Product 5b could be obtained in a preparative scale by treatment of 4b with methanol/0.01% hydrogen chloride for 24 hours at room temperature.

Attempts to synthesize 4a by a 1,3-dipolar cycloaddition of acetonitrile oxide 7 and dihydrotriazolopyrimidine 6 under conditions described for dihydropyridine analogues [1] were unsuccessful; compound 6 remained unchanged.

Compounds 4a,b could principally exist in four diastereomeric pairs of enantiomers. The nmr spectra show a single set of signals. The coupling constants 3J between 3a-H and 4-H in 4a,b amount to 2.4 and 3.0 Hz, respectively. This finding is consistent with an axial-equatorial or with an equatorial-equatorial interaction of the two protons. NOE measurements by irradiation into the signal of 3a-H $(\delta = 3.73 \text{ for 4a} \text{ and 3.39 for 4b})$ confirm the *cis* arrangement of 4-Ph and 3a-H and also the *cis* arrangement of 3a-

H and 9a-CH₃. Thus, coupling constants and NOE results reveal the arrangement shown in Scheme 1 for the three stereogenic centers. Force field calculations (MMX, PCMODEL) [4,5] lead to the same stereochemistry - in so far as the axial orientations of 4-Ph and 9a-CH₃ and the equatorial position of 3a-H correspond to the lowest steric energy [6]. Obviously, hydroxylamine (or hydrazine) attack 3 from the side opposite to the phenyl group. Subsequently a *cis* fused 5-membered ring is formed.

The tricyclic compound 4a represents a novel heterocyclic system, whereas derivatives of 4b are known, which have a carbonyl or a thiocarbonyl group in position C-4 [7-10].

EXPERIMENTAL

The melting points, determined on Kofler apparatus, are uncorrected. The ¹H and ¹³C nmr spectra were obtained on a Bruker AM 400 in deuteriochloroform or dimethyl-d₆ sulfoxide with tetramethylsilane as the internal standard. The mass spectra were recorded on a Finnigan M 95 spectrograph operating at 70 eV.

6-Acetyl-4,7-dihydro-5-methyl-7-phenyl[1,2,4]triazolo-[1,5-a]pyrimidine (3).

A mixture of 0.84 g (10.0 mmoles) of commercially available 3-amino-1,2,4-triazole (1) and 1.88 g (10.0 mmoles) of benzylideneacetylacetone (2) in 1 ml dimethylformamide was refluxed for 0.5 hour. The reaction mixture was cooled to 20°, mixed with 20 ml of benzene and the precipitate was filtered and recrystallized from dimethylformamide. Compound 3 (1.8 g, 71%) was isolated which melted at 230°.

The 1 H nmr signals were found in dimethyl-d₆ sulfoxide at δ 2.13 (s, 3H, 5-CH₃), 2.43 (s, 3H, CH₃CO), 6.45 (s, 1H, 7-H), 7.2-7.3 (m, 5H, ArH), 7.65 (s, 1H, 2-H), 10.7 (bs, 1H, NH).

The 13 C nmr signals were measured in dimethyl-d₆ sulfoxide at δ 19.4 (5-CH₃), 30.4 (6-CH₃), 59.2 (C-7), 107.4 (C-6), 127.2 (o-C_{Ar}), 128.0 (p-C_{Ar}), 128.5 (m-C_{Ar}), 141.5 (i-C_{Ar}), 146.0 (C-2), 146.6 (C-5), 150.0 (C-3a), 194.5 (CO).

The ei mass spectrum had peaks at m/z (%) 254 (45, M^+), 239 (34, M^+ -CH₃), 212 (22), 177 (100, M^+ -C₆H₅).

Anal. Calcd. for C₁₄H₁₄N₄O: C, 66.13; H, 5.55; N, 22.03. Found: C, 66.30; H, 5.60; N, 21.92.

3a,4,9,9a-Tetrahydro-3,9a-dimethyl-4-phenylisoxazolo[5,4-d]-[1,2,4]triazolo[1,5-a]pyrimidine (4a).

A solution of 2.54 g (10.0 mmoles) of 3 and 0.70 g (10.0 mmoles) of hydroxylamine hydrochloride was refluxed in 1 ml pyridine and 50 ml methanol for 12 hours. The mixture was evaporated and the residue crystallized from benzene. Compound 4a (2.0 g, 75%, mp 200-202°) was obtained.

The ¹H nmr spectrum in deuteriochloroform had signals at δ 1.43 (s, 3H, 9a-CH₃), 2.05 (s, 3H, 3-CH₃), 3.73 (d, ³J = 2.4 Hz, 1H, 3a-H), 5.69 (d, ³J = 2.4 Hz, 1H, 4-H), 6.9-7.4 (m, 5H, ArH), 7.56 (s, 1H, 7-H), 7.7 (bs, 1H, NH).

The 13 C nmr signals were measured in deuteriochloroform at δ 12.0 (9a-CH₃), 26.2 (3-CH₃), 56.1 (C-3a), 60.2 (C-4), 93.3 (C-9a), 125.5 (o-C_{Ar}), 128.6 (p-C_{Ar}), 129.2 (m-C_{Ar}), 136.8 (i-C_{Ar}), 149.6 (C-7), 153.6 (C-3), 156.4 (C-8a).

The ei mass spectrum had peaks at m/z (%) 269 (15, M^+), 212 (18, M^+ -CH₃CNO), 197 (10), 186 (43, M^+ -C₂H₃N₄), 171 (37), 135 (100).

Anal. Calcd. for $C_{14}H_{15}N_5O$: C, 62.44; H, 5.61; N, 26.01. Found: C, 62.50; H, 5.71; N, 26.21.

3a,4,9,9a-Tetrahydro-3,9a-dimethyl-4-phenyl-1H-pyrazolo-[3,4-d][1,2,4]triazolo[1,5-a]pyrimidine (4b).

An analogous procedure led to 4b from 3 and hydrazine (without pyridine). The product was isolated in a yield of 80% and melted at 149-151°. The 1 H nmr spectrum in deuteriochloroform had signals at δ 1.31 (s, 3H, 9a-CH₃), 1.93 (s, 3H,

3-CH₃), 3.39 (d, ³J = 3.0 Hz, 1H, 3a-H), 5.65 (d, ³J = 3.0 Hz, 1H, 4-H), 5.7 (bs, 1H, 1-H), 6.9-7.4 (m, ArH), 7.47 (s 1H, 7-H), 7.8 (bs, 9-H).

The 13 C nmr signals were measured in deuteriochloroform at δ 14.7 (9a-CH₃), 26.8 (3-CH₃), 56.8 (C-3a), 59.0 (C-4), 79.0 (C-9a), 126.5 (o-C_{Ar}), 128.7 (p-C_{Ar}), 129.2 (m-C_{Ar}), 138.6 (i-C_{Ar}), 149.3 (C-7), 152.4 (C-3), 155.0 (C-8a).

The ei mass spectrum had peaks at m/z (%) 268 (0.4, M⁺), 185 (100, M⁺ -C₂H₃N₄), 171 (28), 135 (100), 108 (17).

Anal. Calcd. for C₁₄H₁₆N₆: C, 62.67; H, 6.01; N, 31.32. Found: C, 62.82; H, 6.18; N, 31.09.

(5-Amino-1,2,4-triazol-1-yl)(3,5-dimethylpyrazol-4-yl)phenylmethane (5b).

A solution of 0.27 g (1.0 mmole) of **4b** in 5 ml of methanol containing 0.01% hydrogen chloride was allowed to stand for 24 hours at room temperature. The reaction mixture was mixed with 20 ml of water and the precipitate was filtered and recrystallized from benzene. Compound **5b** (0.25 g, 93%) was obtained (mp 191-193°).

The 1 H nmr spectrum in deuteriochloroform had signals at δ 1.91 (s, 6H, 3'-CH₃ and 5'-CH₃), 4.7 (bs, 2H, NH₂), 6.41 (s, 1H, CH), 7.0-7.4 (m, 5H, ArH), 7.50 (s, 1H, 3-H).

The 13 C nmr signals were measured in deuteriochloroform at δ 11.3 (3'-CH₃ and 5'-CH₃), 57.3 (CH), 112.3 (C-4'), 127.1 (o-C_{Ar}), 128.1 (p-C_{Ar}), 128.9 (m-C_{Ar}), 137.7 (i-C_{Ar}), 143.7 (C-3' and C-5'), 148.4 (C-3), 154.2 (C-5).

The ei mass spectrum had peaks at m/z (%) 268 (0.5, M⁺), 185 (100, M⁺ - $C_2H_3N_4$).

Anal. Calcd. for $C_{14}H_{16}N_6$: C, 62.67; H, 6.01; N, 31.32. Found: C, 63.12; H, 5.48; N, 31.40.

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